page 1

Two examples (1 cause-and-effect; 1 [almost certainly] coincidence):

1. **Vinyl chloride** at B.F. Goodrich plant, Louisville, Kentucky and **liver cancer** (angiosarcoma of the liver)



2. **Trichloroethylene** in East Woburn, MA, and **childhood leukemia** (as in, *A Civil Action*)



page 2

How / why are people exposed to vinyl chloride?

Vinyl chloride (monomer) is polymerized to make polyvinyl chloride (PVC) polymer.

Commercial production began in the 1930's.

Worldwide production is enormous:

currently ~ 18 million tons annually.



page 3

Properties of vinyl chloride: Colorless gas; sweet odor. No obvious acute toxicity from early occupational exposures.

Early occupational guideline for allowable exposures (pre-OSHA): 1961: American Council of Government and Industrial Hygienists (ACGIH): TLV of 500 parts per million (ppm, v/v), 8-hr. TWA. Basis?

Research at Dow Chemical (Torkelson *et al.*, **1961**): rodents develop **liver and kidney** toxicity following exposure to 500 ppm; NOAEL = 50 ppm. Recommend reduction in TLV to 50 ppm. Little action taken.

Mid-1960's: Acroosteolysis appears in some Goodrich workers (bone erosion at fingertips). Noticed by plant's occupational physician, John Creech, M.D. (1967). Workers were screened and their hands x-rayed. Some changes made in work procedures (at least at Goodrich) to reduce exposures.

Cancer case reports:

1971: one BF Goodrich (Louisville, KY) VCM worker dies of angiosarcoma of the liver (ASL; rare form of cancer: only about 25 cases per year in U.S. expected).

1973: a second BF Goodrich worker dies of ASL, and another worker is diagnosed with ASL.

Search of company death certificates: a 4th case may have occurred in 1968. (Creech and Johnson [Goodrich's company docs], 1974). **Total workforce at this plant** ~ **300.**

Laura Green

page 4

Case reports of Angiosarcoma of the Liver among Polyvinyl Chloride Workers B.F. Goodrich Plant, Louisville, Kentucky

Case	Age at illness onset	Illness onset	Diagnosis	Death	Years worked with vinyl chloride before illness
1	43	Aug. 1967	Sept. 1967	Jan. 7, 1968	17
2	36	Jan. 1970	May 1970	Sept. 27, 1971	14
3	41	Jan. 1964	Mar. 1973	Mar. 3, 1973	14
4	58	July 1973	Dec. 1973	Dec. 19, 1973	27

February 8, 2005 Environmental chemicals and Cancer clusters: Cause and effect, or coincidence? Laura Green

page 5

Bioassays in lab rodents (Italian researchers):

Rats exposed to **30,000 ppm** VCM develop various tumors (but not ASL: Viola; 1971).

Rats, mice, and hamsters exposed to VCM all develop some ASL and other tumors (Maltoni, early-mid-1970's)

			Number of Animals with Tumors	
Groups	Concentration of vinyl chloride, ppm	No. of Animals (Sprague-Dawley rats)	Liver Angiosarcomas	Nephroblastomas
Ι	200	120	12	3
II	150	120	5	7
III	100	120	1	10
IV	controls	120	0	0
		-		

Rat exposure-response data (Maltoni, 1977)*

*Exposure by inhalation to vinyl chloride in air, at 200, 150, 100 ppm. 4 hr/day, 5 days/week, for 52 weeks (half-lifetime). (Results after 143 weeks = end of the experiments.)

Regulatory responses

Spring 1974: OSHA PEL (which had been **500 ppm**) reduced to an "emergency temporary standard" (ETS) of **50 ppm**, and extensive air monitoring and respiratory protection mandated; ETS superseded within months (January 1, 1975) by a permanent standard of **1 ppm**.

page 6

Industry responses:

Ruling challenged in court -- claimed that PEL was infeasible and exorbitantly expensive (Court disagreed: OHSA won). *But* also (1) Industry sponsored proper epidemiologic studies, (2) U.S.-European VCM-ASL registry established, and (3) Industry-based process engineers quickly re-engineered (enclosed) VC reactors; "hand cleaning" of reactors eliminated; worker exposures *substantially* reduced.

Epidemiologic studies: Early 1980's and 1990's.

Wong *et al.* (1991). ["An industry-wide epidemiologic study of vinyl chloride workers, 1942-1982."] Studied a cohort of 10,173 men who had worked for at least one year in jobs involving exposure to vinyl chloride prior to 1 January 1973. Men employed at 37 plants in the U.S., belonging to 17 companies. A total of 1,536 deaths: 15 deaths from angiosarcoma.

Was industry too slow to respond? Or was it "just" a "different" era?

page 7

May 12, 1959

Mr. W. E. McCormick, Director Department of Industrial Hygiene and Toxicology The **B.F. Goodrich Company** 500 South Main Street Akron, OH

Dear Bill:

This is in reply to your letter of May 4 inquiring about the toxicological information we might have on vinyl chloride.

Insofar as I am aware, there is **no good toxicological data in the literature of the chronic toxicity of vinyl chloride.** As you know, the Conference of Governmental Industrial Hygienists **has for some time been recommending a maximum average of 500 ppm**. This figure is based upon the **acute** data of Patty, *et al.* and, hence, can not be relied upon to strongly when considering chronic exposures. Some years ago, we prepared a little writeup assuming that this figure was alright and a copy of it is enclosed to show you what we have said. I have added my comments to it for your information. I should add that **in the last month we have been investigating vinyl chloride a bit and find it to be somewhat more toxic when given by repeated daily inhalations but it is too early yet to tell what vapor concentrations will be without adverse effect. We feel quite confident, however, that 500 ppm is going to produce rather appreciable injury when inhaled 7 hours a day, five days a week for an extended period.**

As you can appreciate, this opinion is not ready for dissemination yet and I would appreciate it if you would hold it in confidence but use it as you see fit in your own operations. When more data becomes available, I will try to remember to pass it on to you.

Best regards.

Sincerely yours,

V.K. Rowe Biochemical Research Laboratory

February 8, 2005

Environmental chemicals and Cancer clusters: Cause and effect, or coincidence? Laura Green

page 8

Inter-Company Correspondence

Company:	Union Carbide Company	Location: Mel	lon Institute
To: Location:	Dr. T.W. Wale New York Office	Date:	November 24, 1959
Attention Copy To:	Dr. E.R. Weidlein, Jr.		

Dr. A.W. Downes

Dear Tom:

You will recall that the current threshold limit for vinyl chloride is 500 ppm, **based largely on single guinea pig inhalation studies by the Bureau of Mines about 25 years ago**.

An off-the-record phone call from V.K. Rowe gives me incomplete data on their current repeated inhalation study. Six months at 500, 200 and 100 ppm has not found a no-effect level. Even 100 ppm produced organ weight changes and gross pathology, with micropathology expected. Vinyl chloride monomer is more toxic than has been believed. Rowe expects to get more information before he decides whether or not this has any bearing on the safety of packaging uses of vinyl resins.

Dow has been distributed at sales efforts saying chloroform is less toxic than carbon tet. They have completed six months inhalation and find chloroform like carbon tet. This means its threshold limit will be lowered, as I have suggested for years. It was originally set by analogy with carbon tet, and never lowered when that for carbon tet was reduced.

I suggest that these personal communications not be quoted until Dow publishes.

Very truly yours,

Henry F. Smyth, Jr.

February 8, 2005 Environmental chemicals and Cancer clusters: Cause and effect, or coincidence? Laura Green

page 9

What is *A Civil Action* about?

- A. Contaminated drinking water (from municipal wells);
- B. A (space-time) cluster of childhood cancer ("discovered" by a mom and minister); and
- C. The scientific question, "Did A cause B?"
- D. [Other questions: Are such scientific questions best answered *via* lawsuits? Is "justice" thereby served? Who contaminated the water? Is the cluster "statistically significant"? etc...]
- From: Cutler JJ, Parker GS, Rosen S, Prenney B, Healey R, Caldwell GG. 1986. Childhood leukemia in Woburn, Massachusetts. *Public Health Rep.* 101(2):201-5.

"Possible associations between environmental hazards and the occurrence of childhood leukemia were investigated in Woburn, MA, for the period 1969-79. Residents [parents and a minister] of Woburn were concerned over what they perceived to be a large number of childhood leukemia cases . . . Many believed that the elevated rate of childhood leukemia was related to . . . two city water wells that had been closed in 1979 when they were found to be contaminated by organic chemicals. . . This investigation *confirmed an increase in incidence* which was distributed uniformly over the 11-year period. . . . While the contaminants of Wells G and H, which had been closed, are not known leukemogens, it is not possible to rule out exposure to this water as a factor, particularly in the eastern Woburn residents."

BE.104 February 8, 2005

Environmental chemicals and Cancer clusters: Cause and effect, or coincidence? Laura Green

page 10

How sizable / significant was the increased rate? (MA Cancer Registry only established in 1982).

Standardized Incidence Ratios (SIRs) for Leukemia (all types) by Sex in Woburn, Massachusetts January 1969 - December 1983

Sex	Number Observed	Number Expected	Ratio	95% Confidence Limits*
Male	12	3.99	3.00	1.55, 5.25
Female	4	2.02	1.98	0.54, 5.07
Both	19	6.01	3.16	1.90, 4.93

* Fisher exact.

Number of Observed and Expected Cases of Luekemia (all types) by Census Tract in Woburn, Massachusetts January 1969 - December 1983

Number of Cases Census Tract						
	3331	3332	3333	3334	3335	3336
Observed	4	0	1	7	6	1
Expected	1.08	0.82	0.75	.92	1.46	1.00
Poisson Probability	0.02		0.53	5.0 x 10 ⁻⁵	0.004	0.74

From: Cutter (1984)

page 11

Who contaminated the well water in East Woburn?

Plaintiffs' view:

Map removed for copyright reasons.

G and H = Wells G and H

Red Dots = homes of plaintiffs (each with a case of pediatric cancer)

U = Unifirst [Dry-cleaned commercial uniforms, using perc]

C = Cryovac (W.R. **Grace**) [Manufactured food processing and packaging equipment, 1960-1988; used TCE]

R = Riley Tannery [**1910-**1990's: property purchased by **Beatrice** Foods in 1978 (contamination discovered in 1979)]

page 12

Who contaminated the water?

Map removed for copyright reasons.

U.S. EPA view:

probably more complicated than this.



February 8, 2005 Environmental chemicals and Cancer clusters: Cause and effect, or coincidence? Laura Green

page 13

Woburn is an old city (incorporated 1642)

1800's: Manufacturing center, tanning leather, making shoes and boots.

1865: "Woburn was at the head of the tanning industry in the country."

1884: 26 large tanneries operating.

1901: Henry Thayer of Woburn originates chrome (Cr VI) tanning (replaces bark tanning)

By 1915: Some diversification: making ice cream, machine tools, mops, and paper boxes.

1920s: Groundwater wells for drinking water, the first in Massachusetts, began operating. [Eventually, 6 wells (A - F) drilled into the groundwater aquifer surrounding **Horn Pond** (south-central Woburn)]

Early 1960's: Population growing, water becoming scarce: city officials consider drilling wells in groundwater-rich, swampy, industrialized East Woburn.

1964: Drinking water Well G constructed (along the east floodplain of the Aberjona River)

1967: Well H constructed. Soon thereafter, recommendations to take wells G & H out of service due to bacterial contamination: heavily chlorinated instead. Resident: "The odor is almost like a clear bleach"

Early-mid 1970's: the State warns that water in Wells G & H is of poor quality, with elevated levels of nitrates, chlorides, sulfates, sodium, manganese, hardness, and elevated levels of carbon-chloroform extract [crude indication of organic contaminants]

1979: TCE and perc detected

February 8, 2005 Environmental chemicals and Cancer clusters: Cause and effect, or coincidence? Laura Green

page 14

What are our sources of drinking water in Massachusetts?

- Groundwater wells public and private.
- Surface water from reservoirs or rivers.

2004 Public Drinking Water System Reliance on Ground Water and Surface Water Sources in New England

Ground Water

Surface Water

State	Population	% of Population	Population	% of Population
Connecticut	462,052	17	2,231,612	83
Massachusetts	2,065,285	43	2,709,203	57
Maine	453,058	52	417,863	48
New Hampshire	595,129	56	474,976	44
Rhode Island	209,548	28	551,162	72
Vermont	345,659	56	266,510	44

page 15

What was the primary waterborne contaminant? (What about the *other* contaminants, chemical and microbiological . . .?)

Trichloroethylene (TCE)

Colorless, volatile liquid; Versatile organic solvent Reasonably chemically stable Very low flammability: *no measurable flashpoint* Not very toxic acutely; Not unpleasant odor

Widely used in industry (since about 1930)metal cleaning / degreasing55 %chemical intermediate41 %

Historically, use of TCE replaced use of CCl_4 and $CHCl_3$, which had replaced use of petroleum distillates.

Concentration in Woburn water $\approx 260 \ \mu g / \text{liter} (260 \text{ ppb})$ U.S. EPA "Maximum Contaminant Level" (MCL) = 5 ppb. (TCE is a very common groundwater, hence drinking water, contaminant)

Does gross "exceedance" of MCL mean that children (fetuses) exposed to this water developed cancer as a result?

page 16

Leukemias: 4 basic types: lymphoid and myeloid (acute and chronic). All arise from cells in the **bone marrow** (soft inner part of some bones)

Three diagrams removed for copyright reasons.

page 17



What do we know about acute lymphoblastic leukemia (ALL)?

The horizontal axis shows 5-year age intervals. The vertical axis shows the frequency of new cases of ALL per 100,000 in a given age group. Note that the risk of ALL is greatest in the first 5 years of life. (Data from SEER Program of the National Cancer Institute.)

ALL is the most common of all **pediatric** cancers: $\approx 2,000$ new cases of childhood ALL diagnosed annually in U.S.

Cause(s)? Unknown. Timing? "Mistakes" during fetal development? Exposures *in utero*? (Offspring of cigarette smoking moms: are they at excess risk of ALL?).

Possible causes:

Bad luck Genetic predisposition (e.g., Down syndrome) (Retro)virus(es)? (Analogy with other mammals) EMF? Woburn: contaminated drinking water???

February 8, 2005 Environmental chemicals and Cancer clusters: Cause and effect, or coincidence? Laura Green

page 18

What do we know about cancer clusters *in communities*?

From: Caldwell GG. 1990. Twenty-two years of cancer cluster investigations at the Centers for Disease Control. Am J Epidemiol. 132(1 Suppl):S43-7:

Beginning in 1961 [and spanning 20 years of investigation], the Centers for Disease Control investigated 108 cancer clusters . . . The clusters studied were of leukemia (38%), leukemia and lymphoma (30%), leukemia and other cancer combinations (13%), and all other cancer or combinations (19%). . . . Although 14 different categories of associations were reported [and despite lab tests for radiation, chemicals, viruses, and genetics], *no clear cause* was found for *any* cluster.

- A reminder that, in *community (chronic) disease investigations, most* hypotheses are *wrong*.
- Does TCE cause cancer in over-exposed factory workers? Lots of **epidemiologic** study of this question.
- Answer: *Probably* not. . . "limited evidence." No "unusual" cancers. TCE not IARC "Group 1."

February 8, 2005

Environmental chemicals and Cancer clusters: Cause and effect, or coincidence? Laura Green

page 19

Overall Evaluations of Carcinogenicity to Humans (per IARC Monographs Volumes 1-88) Group 1: Carcinogenic to humans (95): (Established via sufficient epidemiologic evidence)

AGENTS AND GROUPS OF AGENTS	Human papillomavirus type 16
Aflatoxins	Human papillomavirus type 18
4-Aminobiphenyl	Human T-cell lymphotropic virus type I
Arsenic and arsenic compounds	Melphalan
Asbestos	8-Methoxypsoralen (Methoxsalen) plus
Azathioprine	ultraviolet A radiation
Benzene	MOPP and other combined chemotherapy
Benzidine	including alkylating agents
Beryllium	Mustard gas (Sulfur mustard)
N,N-Bis(2-chloroethyl)-2-naphthylamine	2-Naphthylamine
(Chlornaphazine)	Neutrons
Bis(chloromethyl)ether and chloromethyl	Nickel compounds
methyl ether	Opisthorchis viverrini
1,4-Butanediol dimethanesulfonate	Oral contraceptives, combined
(Busulphan; Myleran)	Oral contraceptives, sequential
Cadmium and cadmium compounds	Phosphorus-32, as phosphate
Chlorambucil	Plutonium-239 and its decay products as
1-(2-Chloroethyl)-3-(4-methylcyclohexyl)-1-ni	aerosols
trosourea (Methyl-CCNU; Semustine)	Radioiodines, short-lived isotopes, including
Chromium[VI] compounds	iodine-131, from atomic reactor accidents and
Ciclosporin	nuclear weapons detonation
Cyclophosphamide	Radionuclides, a-particle-emitting, internally
Diethylstilboestrol	deposited
Epstein-Barr virus	Radionuclides, b-particle-emitting, internally
Erionite	deposited
Estrogen therapy, postmenopausal	Radium-224 and its decay products
Estrogens, nonsteroidal	Radium-226 and its decay products
Ethylene oxide	Radium-228 and its decay products
Etoposide in combination with cisplatin and	Radon-222 and its decay products
bleomycin	Schistosoma haematobium
Formaldehyde	Silica crystalline
[Gamma Radiation: see X- and Gamma	Solar radiation
(g)-Radiation]	Talc containing asbestiform fibres
Gallium arsenide	Tamoxifen
Helicobacter pylori	
Hepatitis B virus	
Hepatitis C virus	
Herbal remedies containing plant species of	
the genus Aristolochia	
Human immunodeficiency virus type 1	

page 20

Group 1: Carcinogenic to humans (95), continued:

2,3,7,8-Tetrachlorodibenzo-*p*-dioxin Thiotepa

Thorium-232 and its decay products Treosulfan Vinyl chloride X- and Gamma (g)-Radiation

MIXTURES

Alcoholic beverages Analgesic mixtures containing phenacetin Areca nut Betel quid with tobacco Betel quid without tobacco Coal-tar pitches Coal-tars Mineral oils, untreated and mildly treated Salted fish (Chinese-style) Shale-oils Soots Tobacco products, smokeless Wood dust

EXPOSURE CIRCUMSTANCES Aluminium production Arsenic in drinking-water Auramine, manufacture of Boot and shoe manufacture and repair Coal gasification Coke production Furniture and cabinet making Haematite mining (underground) with exposure to radon Involuntary smoking Iron and steel founding Isopropanol manufacture Magenta, manufacture of Painter (occupational exposure as a) Rubber industry Strong-inorganic-acid mists containing sulfuric acid (occupational exposure to) **Tobacco** smoking

page 21

What about results in lab rats and mice exposed chronically to TCE?

Table Doses of trichloroethylene (TCE): toxicologic and other benchmarks				
Endpoint	TCE (mg/kg-day)			
Doses to people from water in Woburn	??			
Worker's day-long exposure to CalOSHA PEL (25 ppm)	??			
Mouse NOAEL: liver cancer (half-lifetime exposure; Herren-Freund <i>et al.</i> , 1987)	6			
Rat NOAEL: kidney cancer (one-year exposure; Maltoni <i>et al.</i> , 1986)	250			
Rat LOAEL: kidney cancer (NTP, 1988)	500			
Mouse LOAEL: liver cancer (NTP, 1990)	1,000			

page 22

TCE (and other solvents) elsewhere in groundwater in the U.S.

1980: Comprehensive Environmental Response, Compensation, and Liability Act (CERCLA;), commonly known as "Superfund" enacted: money from a tax on the chemical and petroleum industries, mandates U.S. EPA "respond directly to releases or threatened releases of hazardous substances that *may* endanger public health or the environment." > 1,400 National Priorities List (NPL) sites in U.S.; TCE has been detected at some 850 of these!



page 23

Per the U.S. Agency for Toxic Substances and Disease Registry (ATSDR)

"Drinking or breathing high levels of trichloroethylene may cause nervous system effects, liver and lung damage, abnormal heartbeat, coma, and possibly death."

Are statements like this informative? [ATSDR's motto: "To Protect America's Health from Toxic Exposures"]

What concentration of TCE in groundwater that serves (or may serve) as drinking water is *acceptably small*? (Aquifer clean up = "remediation" = improving public health?)

Quantitative health risk assessment for TCE: nothing (really) to do with Woburn. Based instead on results of **rat/mouse bioassays**, combined with **massive regulatory infrastructure**...

Huge \$\$ at stake.

TCE toxicology, mechanisms, and QRA for another time. . .

page 24

For now, by next Tuesday, Feb. 15, 2005, please do the following and email to me your answers:

Compare and contrast *former occupational* exposures to vinyl chloride in air with *drinking water* exposures to TCE in Woburn (and *currently permissible workplace exposures* to TCE). Express doses as mg/kg-day, with plausible ranges.

In air, X ppm = (Y mg/m3)(24.45)/(molecular weight)

In water, $\mathbf{x} \text{ ppb} = \mathbf{x} ug/L$