

# Lipids

- Molecules of Mr =150 -  $\approx$  2000 composed of saturated, unsaturated and/or aromatic or aliphatic hydrocarbon moieties - *Non-polar lipids*
- When water-binding functional groups (-OH, -COOH, -NH, -C=O, etc.) are covalently linked - *Polar Lipids*
- Biologically-relevant lipids are molecules with aliphatic chains of at least 12C atoms and/or aromatic/aliphatic structures with at least 3 rings which may be fused
- Old system of classification based on solubility in organic solvents is neither strictly true nor useful (e.g., bile salts)

# OCTADECANOL

Graphic representations of a Polar  
Lipid

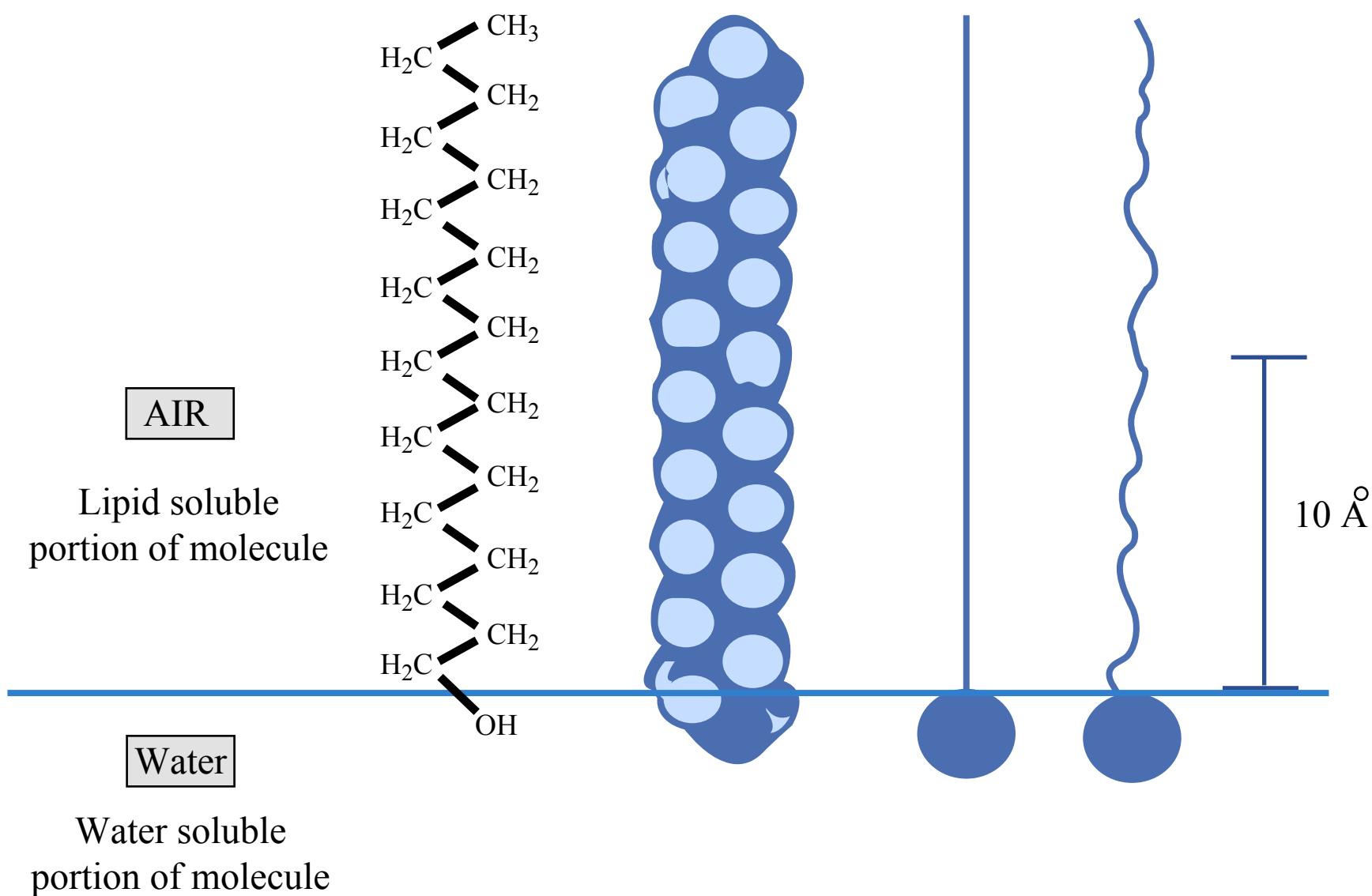


Figure by MIT OCW.

# Classification of Polar Lipids Based on Interactions with H<sub>2</sub>O\*

\*D. M. Small (1968)

## Class

## Surface and Bulk Interactions with Water

### Non-Polar Lipids

Will Not Spread To Form A Monolayer  
Insoluble In Bulk

### Polar Lipids

#### A. Insoluble Non-Swelling Amphiphiles

Forms A Stable Monolayer Insoluble In Bulk

#### B. Insoluble Swelling Amphiphiles

Forms A Stable Monolayer  
Bulk Phase-pure liquid crystals in pure water

#### C. Soluble Amphiphiles

##### 1. with lyotropic mesomorphism



→ L.C. → Micelle

Forms An Unstable Monolayer  
Bulk Phase-a micellar solution

##### 2. without lyotropic mesomorphism



→ Micelle

Forms An Unstable Monolayer  
Bulk Phase-a micellar solution

# Self-Aggregated States

## i) LIQUID CRYSTALS (L.C.)

- Intermediate Physical States (mesophases) with properties of both liquids and solid crystals.
- Long range order in at least 1 dimension
  - Lyotropic L.C.
  - Thermotropic L.C.
- Have distinct optical textures by polarizing microscopy

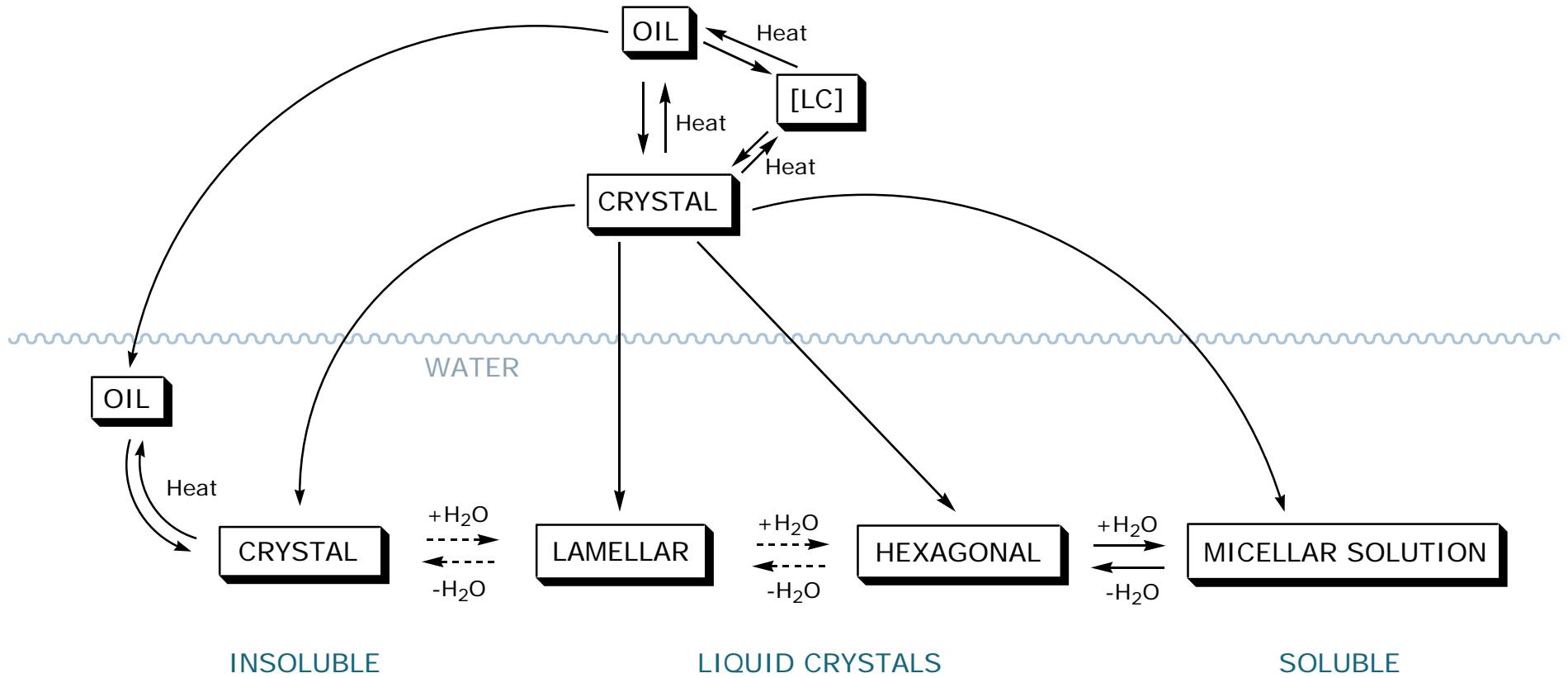


Figure by MIT OCW.

# Self Aggregated States

## ii) MICELLES

Thermodynamically stable aggregates of soluble amphiphilic lipids that form spontaneously above a critical micellar concentration (CMC) and critical micellar temperature (CMT)

- in aqueous systems: regular micelles  
*“The hydrophobic effect”*
- in organic solvents: reverse micelles

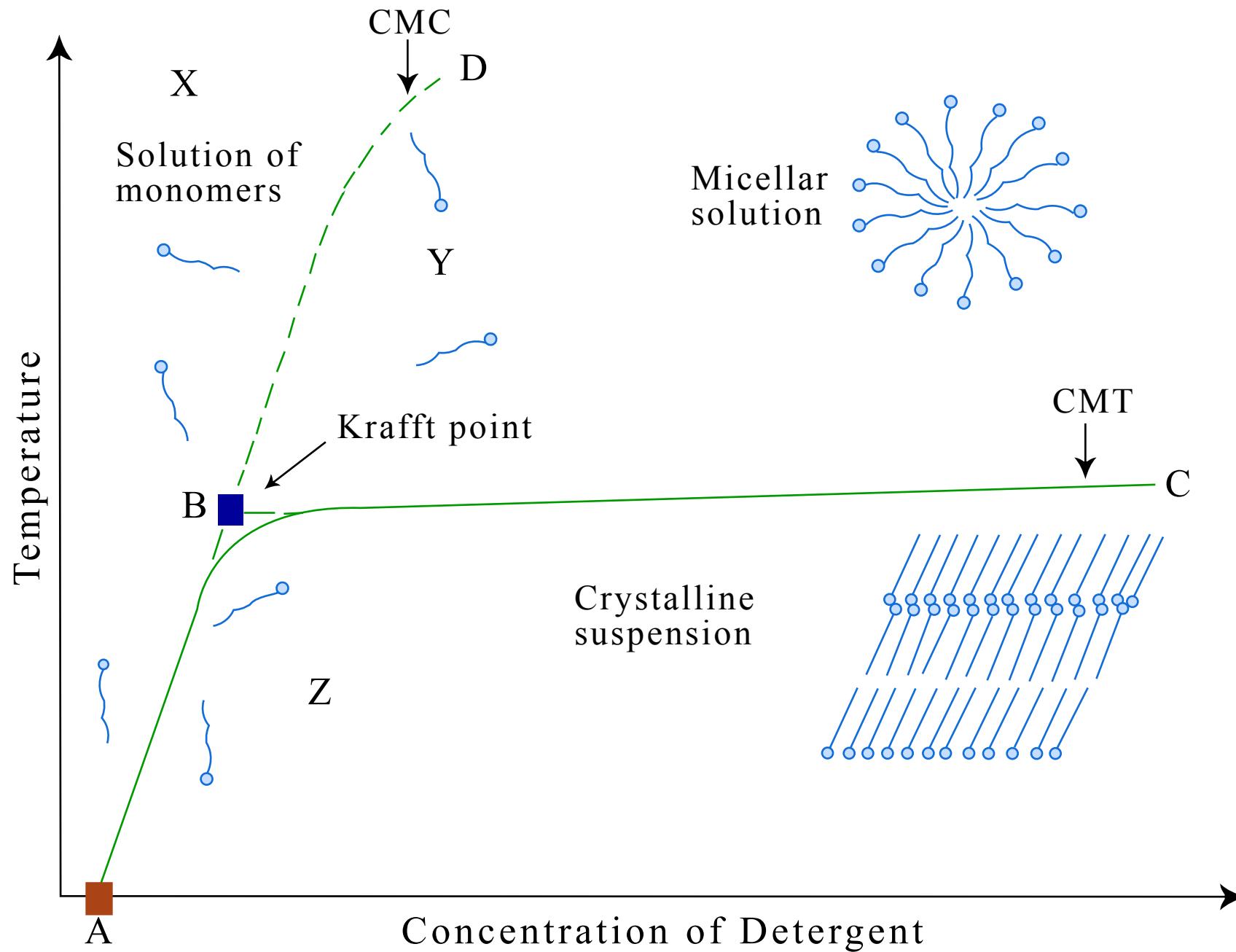


Figure by MIT OCW.

# Self-Aggregated States

## iii) EMULSIONS

Dispersions of one liquid in a continuous phase of another liquid: O/W, W/O systems

The dispersed (discontinuous) phase consists of microscopic droplets, usually 0.1-100  $\mu\text{m}$  in diameters

# Self-Aggregated States

## iv) SOLID CRYSTALS

Classically lipids such as Cholesterol (Gallstones, Atheroma), fatty acids + bile acids (Enteroliths), glycolipids (neural storage diseases) – all are pathologic

# Anhydrous Cholesterol and Cholesterol Monohydrate

Figures removed due to copyright reasons. Please see:

Shieh, H. S., et al. "Crystal structure of anhydrous cholesterol." *Nature* 267 (1977): 287-9.

Craven, B. M. "Crystal structure of cholesterol monohydrate." *Nature* 260 (1976): 727-9.

Figure 1 in Loomis, C. R., et al. "The phase behavior of hydrated cholesterol." *J Lipid Res* 20 (1979): 525-535.

# Principal Mixed Lipid Systems in Living Organisms

- Stable Emulsions (dietary fat, plasma lipoproteins, intracellular fat droplets, gut luminal lipids pre-digestion, etc.)
- Mixed Micelles (bile, gut lumen, certain brain lipid storage diseases)
- Mixed Liquid Crystals (biologic membranes, serum lipoprotein X in cholestasis, myelin sheet, mixed vesicles in gut lumen, etc.)

# BEHAVIOR OF LECITHIN IN WATER

Figure removed due to copyright reasons.

## ADDITION OF BILE SALT TO LECITHIN - CHOLESTEROL LIQUID CRYSTAL

Figure removed due to copyright reasons.

# The 3 “P” Rules

- Predictability Rule
- Predominance Rule
- Phase Rule ( $F=C-P+2$ )

# How Lipids Traverse Biological Membranes

- As single molecules (molecule need not be water soluble)
- As aggregated particles (i.e., stable emulsions)
- Transporter control: Genomic (slow), nongenomic (fast)

## CELLULAR CHOLESTEROL HOMEOSTASIS

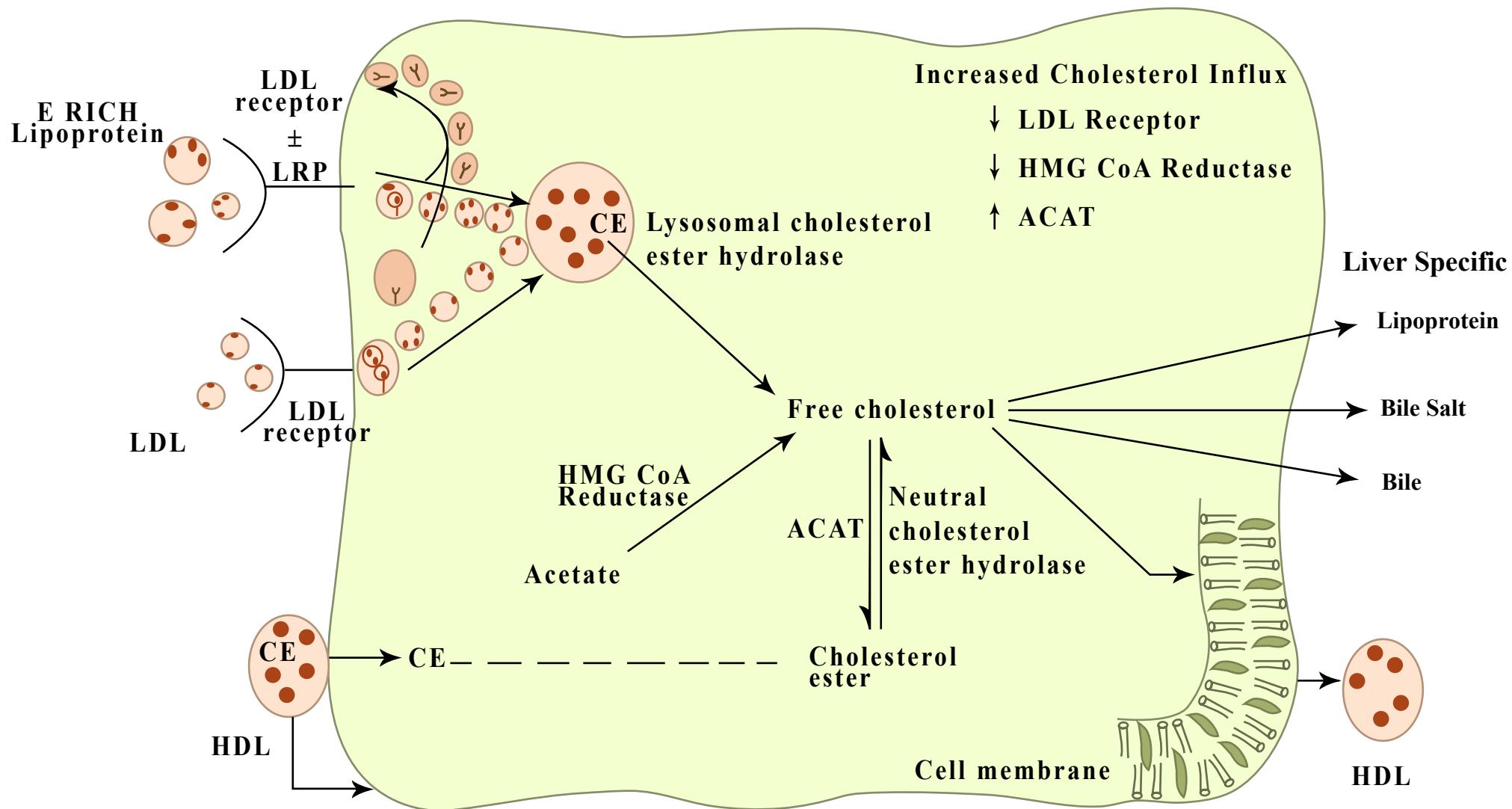
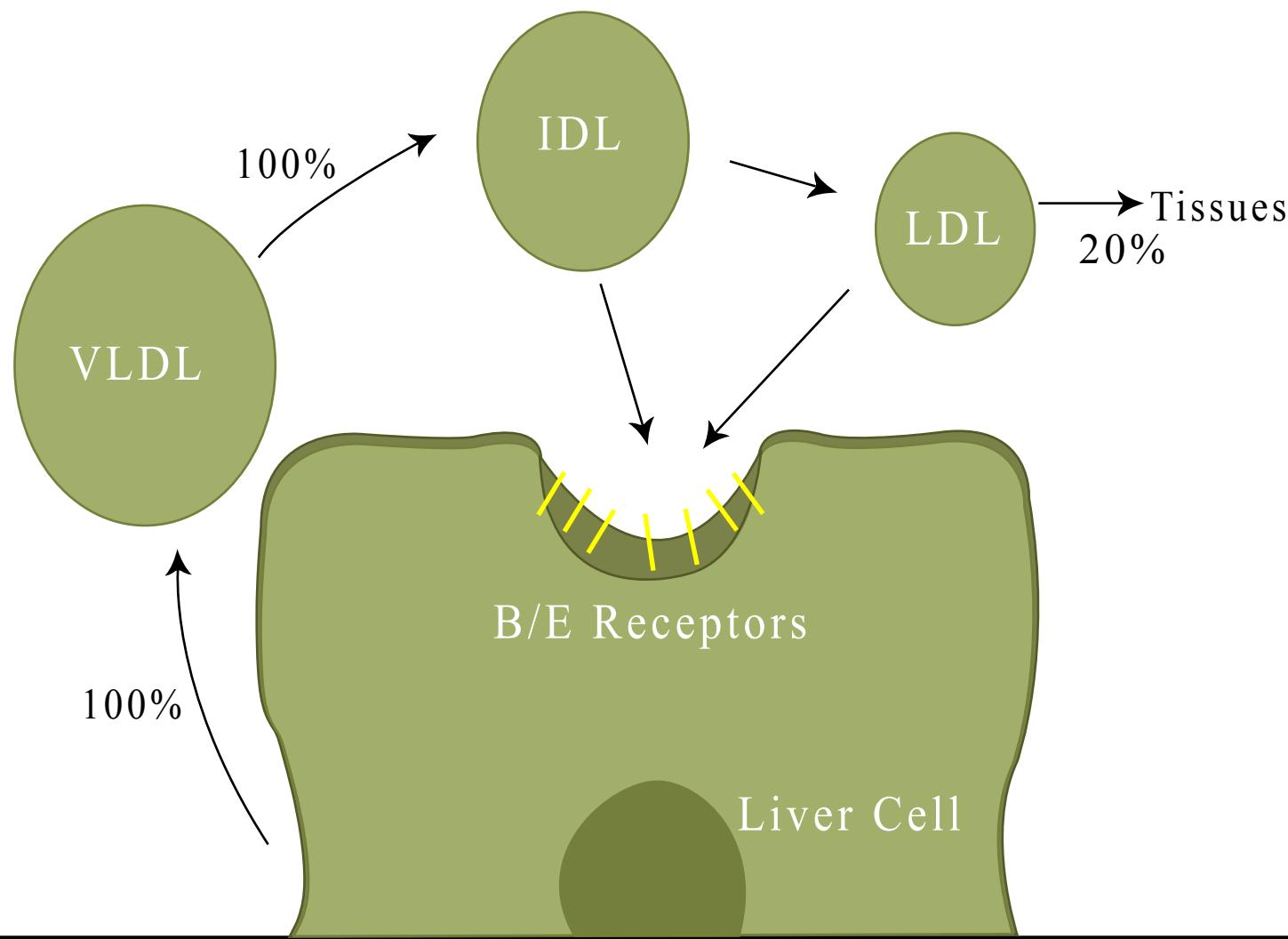


Figure by MIT OCW.

## The Liver And Plasma Lipid Homeostasis

1. All VLDL made in liver.
2. ~ 80% of LDL removed by liver.



# CHEMICAL COMPOSITIONS OF HUMAN PLASMA LIPOPROTEINS

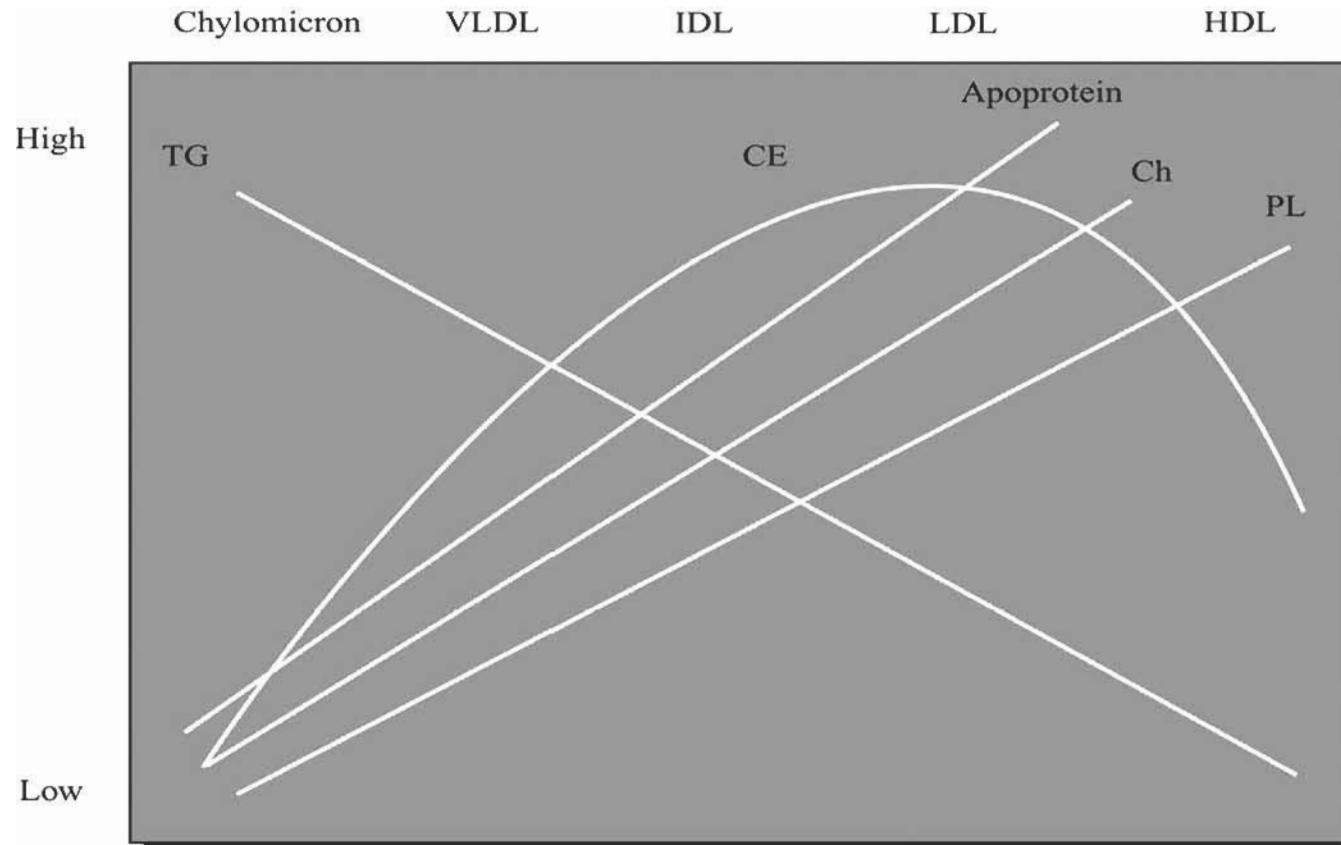


Figure by MIT OCW.

# GI Movements of Single Molecules: Examples

	<u>Bile Salts</u> (soluble)	<u>Cholesterol/Phytosterols</u> (insoluble)
<b><u>Enterocytes:</u></b>	Distal Ileocytes  Influx: ASBT, FABP6, OAT $\alpha/\beta$	Proximal > Distal  Influx: NPC1L1  Efflux: ABCG5/ABCG8
<b><u>Transport:</u></b>	Portal Blood	Lymphatics
<b><u>Binding:</u></b>	Albumen, HDL	ChE and free Ch in chylomicrons and nascent HDL
<b><u>Hepatic Uptake:</u></b>	NTCP – 80%  OATPs – 20%	ApoB/E receptor  LRP receptor
<b><u>Nuclear Control:</u></b>	FXR/RXR  SHP, LRH1	SREBP's  LXR/RXR
<b><u>Biliary Secretion:</u></b>	BSEP, MRP2	ABCG5/ABCG8, (others unknown)
<b><u>Facilitators:</u></b>	Needs intact FIC1 function	Same + hydrophobic “sink” in bile

**That's All Folks!**