## Problem 5.4

Define points on 20C line

> with(geometry):point(p1\_20,44e6/(20+273),ln(1e-4)): > point(p2\_20,53.2e6/(20+273),ln(1e-2)):

get slope of line and multiply by 2R to get activation volume

> Digits:=4:'V (m^3/mol)'= 2\*8.314\*slope(p2\_20,p1\_20);

$$V\left(\frac{m^3}{mol}\right) = .002440$$

Compute activation energy by horizontal difference between 20C and -60C lines > eq:=(V/2)\*((69e6/(-60+273)) -

(44e6/(20+273)))=Delta[H]\*(1/(-60+273) - 1/(20+273));

$$eq := 211.9 = \frac{80}{62409} \Delta_{H}$$

> 'Delta[H] (kJ/mol) ' = solve(eq,Delta[H])/1000;

$$\Delta_{H}\!\!\left(\frac{kJ}{mol}\right) = 165.3$$