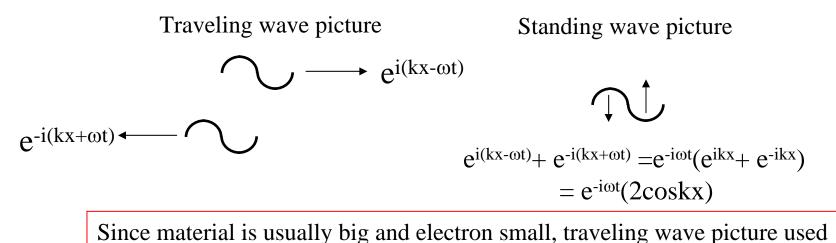
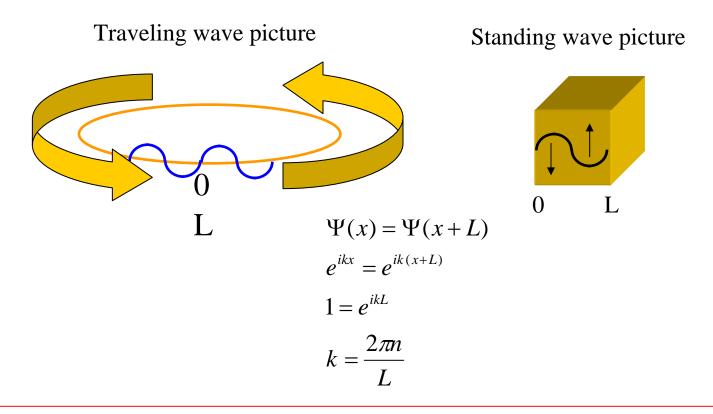
Consequence of Electrons as Waves on Free Electron Model

- Boundary conditions will produce quantized energies for all free electrons in the material
- Two electrons with same spin can not occupy same electron energy (Pauli exclusion principle)

Imagine 1-D crystal for now

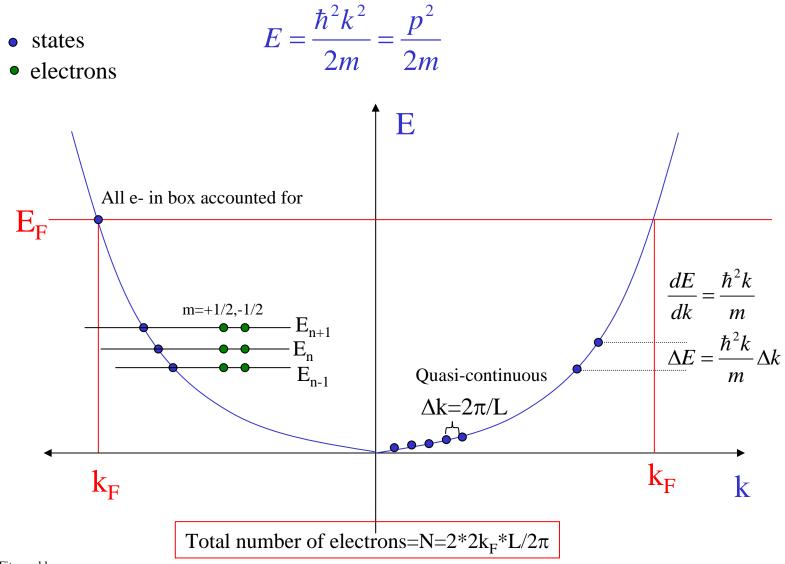


Consequence of Electrons as Waves on Free Electron Model

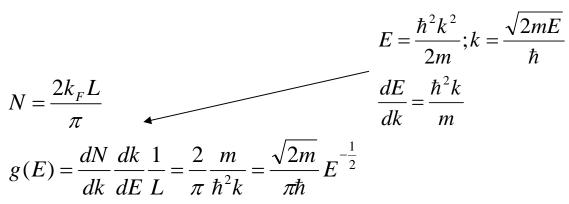


Just having a boundary condition means that k and E are quasi-continuous, i.e. for large L, they appear continuous but are discrete

Representation of E,k for 1-D Material



Representation of E,k for 1-D Material

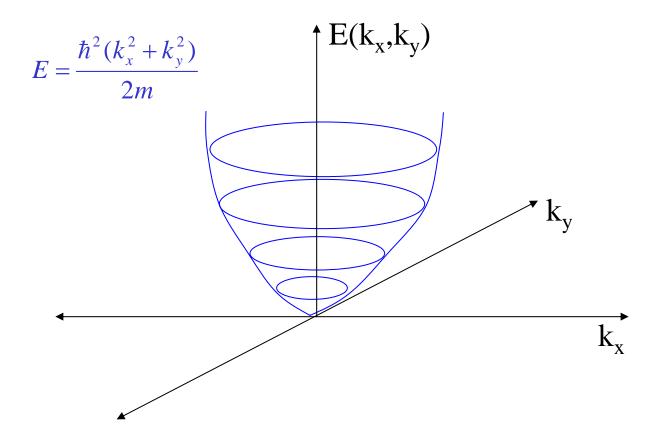


g(E)=density of states=number of electrons per energy per length

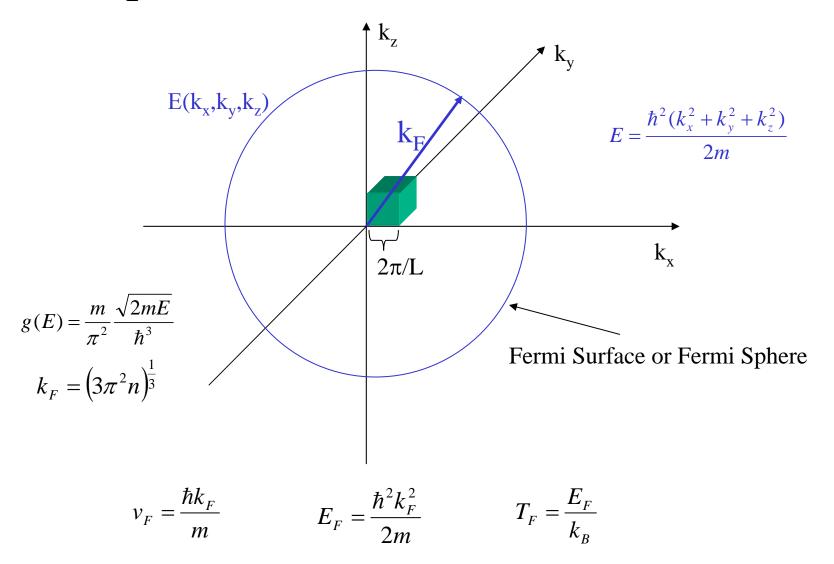
$$n = \frac{N}{L} = \frac{2k_F}{\pi} = \frac{2\sqrt{2mE_F}}{\hbar\pi} \text{ or } k_F = \frac{n\pi}{2}$$

n=is the number of electrons per unit length, and is determined by the crystal structure and valence
The electron density, n, determines the energy and velocity of the highest occupied electron state at T=0

Representation of E,k for 2-D Material



Representation of E,k for 3-D Material



So how have material properties changed?

- The Fermi velocity is much higher than kT even at T=0! Pauli Exclusion raises the energy of the electrons since only 2 e- allowed in each level
- Only electrons near Fermi surface can interact, i.e. absorb energy and contribute to properties

$$T_{F} \sim 10^{4} K \ (T_{room} \sim 10^{2} K),$$

 $E_{F} \sim 100 E_{class}, v_{F}^{2} \sim 100 v_{class}^{2}$

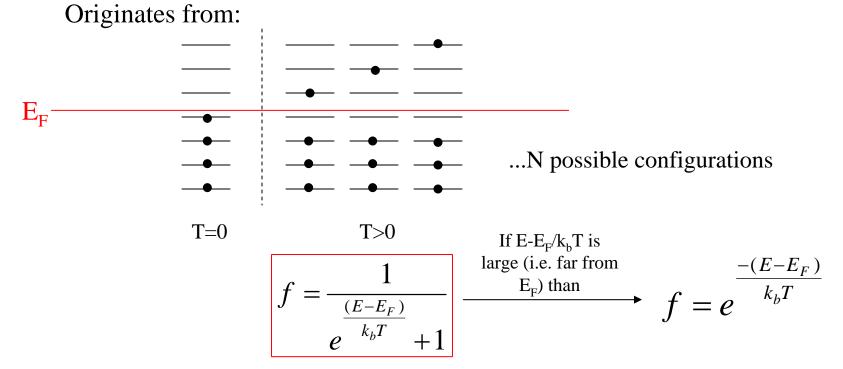
Element	r_s/a_0	ϵ_F	T_F	k_F	v_F
Li	3.25	4.74 eV	5.51 x 10 ⁴ K	1.12 x 10 ⁸ cm ⁻¹	1.29 x 10 ⁸ cm/se
Na	3.93	3.24	3.77	0.92	1.07
K	4.86	2.12	2.46	0.75	0.86
Rb	5.20	1.85	2.15	0.70	0.81
Cs	5.62	1.59	1.84	0.65	0.75
Cu	2.67	7.00	8.16	1.36	1.57
Ag	3.02	5.49	6.38	1.20	1.39
Au	3.01	5.53	6.42	1.21	1.40
Be	1.87	14.3	16.6	1.94	2.25
Mg	2.66	7.08	8.23	1.36	1.58
Ca	3.27	4.69	5.44	1.11	1.28
Sr	3.57	3.93	4.57	1.02	1.18
Ba	3.71	3.64	4.23	0.98	1.13
Nb	3.07	5.32	6.18	1.18	1.37
Fe	2.12	11.1	13.0	1.71	1.98
Mn	2.14	10.9	12.7	1.70	1.96
Zn	2.30	9.47	11.0	1.58	1.83
Cd	2.59	7.47	8.68	1.40	1.62
Hg	2.65	7.13	8.29	1.37	1.58
Al	2.07	11.7	13.6	1.75	2.03
Ga	2.19	10.4	12.1	1.66	1.92
In	2.41	8.63	10.0	1.51	1.74
Tl	2.48	8.15	9.46	1.46	1.69
Sn	2.22	10.2	11.8	1.64	1.90
Pb	2.30	9.47	11.0	1.58	1.83
Bi	2.25	9.90	11.5	1.61	1.87
Sb	2.14	10.9	12.7	1.70	1.96

Fermi energies, fermi temperatures, fermi waves vectors, and fermi velocities for representative metals^{*} * The table entries are calculated from the values of r_s / a_0 given in Table 1.1 using $m = 9.11 \times 10^{-28}$ grams.

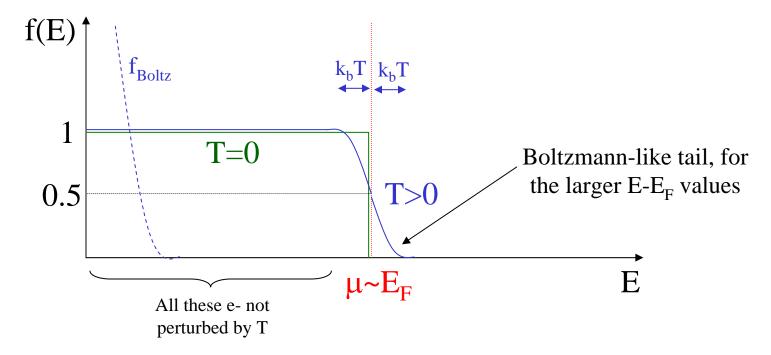
Table by MIT OpenCourseWare.

Effect of Temperature (T>0): Coupled electronic-thermal properties in conductors (i.e. c_v)

- Electrons at the Fermi surface are able to increase energy: responsible for properties
- Fermi-Dirac distribution
- NOT Bolltzmann distribution, in which any number of particles can occupy each energy state/level



Fermi-Dirac Distribution: the Fermi Surface when T>0



Heat capacity of metal (which is ~ heat capacity of free e- in a metal):

U=total energy of alastrong in gustam

$$c_{v} = \left(\frac{\partial U}{\partial T}\right)_{v} \qquad U \sim \Delta E \cdot \Delta N \sim k_{b}T \cdot \left[g(E_{F}) \cdot k_{b}T\right] \sim g(E_{F}) \cdot \left(k_{b}T\right)^{2}$$

$$c_{v} = \left(\frac{\partial U}{\partial T}\right)_{v} = 2 \cdot g(E_{F}) \cdot k_{b}^{2}T \qquad \text{Right dependence, very close to exact derivation}$$

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Heat Capacity (c_v) of electrons in Metal

- Rough derivation shows $c_v \sim \text{const. x T}$, thereby giving correct dependence
- New heat capacity is about 100 times less than the classical expectation

Exact derivation:
$$c_v = \frac{\pi^2}{3} \cdot k_b^2 T \cdot g(E_F)$$

$$\frac{c_{vclass}}{c_{vquant}} = \frac{\frac{5}{2}nk_b}{\frac{\pi^2}{2}\left(\frac{k_bT}{E_F}\right)nk_b} = \frac{3}{\pi^2}\frac{E_F}{k_bT} \sim 100 @ RT$$